

DRY ICE

PHYSICAL & CHEMICAL PROPERTIES

Carbon dioxide (chemical symbol = CO₂) is a natural by-product of respiration, fermentation and other industrial processes. It is in the form of a colorless, odorless and non-flammable gas. Its density at 25°C is 1.98 kg/m³ (0.123 lb/ft³ at 32°F), about 1.65 times that of air, i.e. CO₂ gas displaces oxygen in the environment (the CO₂ sinks under the oxygen).

Dry ice which corresponds to the solid form of CO₂ has an opaque white appearance.

TECHNICAL INFORMATION

Purity :	Impurities :	
CO ₂	Oil	CO
≥ 99 %	5mg / kg in liquide CO ₂	≤ 10 ppm

- EC Regulations 178/2002, 852/2004 and 1935/2004
- EC Directives 96/77, 2002/72
- EIGA recommendations: IGC doc 150/18



APPLICATIONS :

The use of dry ice is as extensive as it is varied.

- 1- Scientific and medical research:** useful for the preservation of organs and biological products, the cooling of exothermic reactions.
- 2- In restaurants and haute cuisine:** In this sector, the use of dry ice opens up a world full of opportunities to create astonishing and attractive dishes for the public who consume them.
- 3- Industry:** In this sector, it is used to facilitate the assembly and adjustment of parts by cold shrinkage, but also for cryogenic grinding or deburring of molded plastics and rubbers.
- 4- Computer and electronics sector:** Cooling electronic devices (overclocking) is very interesting for increasing their performance, by intensifying the diffusion of electrical signals.
- 5- Transport:** dry ice is used for shipping frozen or deep-frozen products.
- 6- Cryogenic cleaning:** Use of dry ice in the form of pellets by projection.
- 7- For entertainment:** dry ice is the ideal tool for producing smoke during events.