



Molar mass: 31.9988 g/mol

Melting point: -219°C Boiling point: -183°C

Density of the gas phase (1.013 bar and

15°C): 1.354 kg/ma

Density of the liquid phase (1.013 bar at

boiling point): 1.1415 kg/l

Gas density (1.013 bar at boiling point):

4.475 kg/m

Latent heat of fusion (1.013 bar at triple point): 13.9 kJ/kg

Latent heat of vaporization (at 1.013 bar

boiling point): 212.98kJ/kg Critical temperature: -118.6°C Critical pressure: 50.43 bar

Compressibility factor (Z) (1.013 bar and

15°C): 0.994

Concentration in air: 20.94% vol.

APPLICATIONS:

Health: Assisted breathing

A vital distress

Poisoning from fire smoke or carbon monox-

ide

A decompression accident secondary to a dive or work in a hyperpressure environment (tunnel boring machines).

A SpO² measurement that indicates a value <94%

A painful crisis in a victim who has a history of sickle cell anemia.

Laboratories and analyses: Oxygen is used to calibrate trace impurity analyzers, safety detectors, environmental control analyzers, working atmosphere control analyzers.

TECHNICAL INFORMATION

Purity:	Impurities :			
02	CO	CO2	H2O	
≥ 99.5%	≤ 5 ppm	≤ 300 ppm	≤ 67 ppm	

Conditioning:

B2	B5	B10	B50
$0.5 \mathrm{m}^{3}$	1 m³	1.5 m ³	8 m³

European Pharmacopoeia

